

## Appendix B - Properties and Toxicities of Organic Solvents

The values in the table below except as noted have been extracted from online and hardbound compilations. Values for relative polarity, eluant strength, threshold limits and vapor pressure have been extracted from: Christian Reichardt, *Solvents and Solvent Effects in Organic Chemistry*, Wiley-VCH Publishers, 3rd ed., 2003.

**TABLE 1**

<u>Solvent</u>	formula	boiling point (°C)	melting point (°C)	density (g/mL)	solubility in H <sub>2</sub> O <sup>1</sup> (g/100g)	relative polarity <sup>2</sup>	eluant strength <sup>3</sup>	threshold limits <sup>4</sup> (ppm)	vapor pressure 20°C (hPa)
acetic acid	C <sub>2</sub> H <sub>4</sub> O <sub>2</sub>	118	16.6	1.049	M	0.648	>1	10	15.3
acetone	C <sub>3</sub> H <sub>6</sub> O	56.2	-94.3	0.786	M	0.355	0.56	500	240
acetonitrile	C <sub>2</sub> H <sub>3</sub> N	81.6	-46	0.786	M	0.460	0.65	20	97
acetyl acetone	C <sub>5</sub> H <sub>8</sub> O <sub>2</sub>	140.4	-23	0.975	16	0.571			
2-aminoethanol	C <sub>2</sub> H <sub>7</sub> NO	170.9	10.5	1.018	M	0.651		3	0.53
aniline	C <sub>6</sub> H <sub>7</sub> N	184.4	-6.0	1.022	3.4	0.420		2	0.4
anisole	C <sub>7</sub> H <sub>8</sub> O	153.7	-37.5	0.996	0.10	0.198			
benzene	C <sub>6</sub> H <sub>6</sub>	80.1	5.5	0.879	0.18	0.111	0.32	0.5	101
benzonitrile	C <sub>7</sub> H <sub>5</sub> N	205	-13	0.996	0.2	0.333		10	12
benzyl alcohol	C <sub>7</sub> H <sub>8</sub> O	205.4	-15.3	1.042	3.5	0.608			
1-butanol	C <sub>4</sub> H <sub>10</sub> O	117.6	-89.5	0.81	7.7	0.586		20	6.3
2-butanol	C <sub>4</sub> H <sub>10</sub> O	99.5	-114.7	0.808	18.1	0.506		100	
<i>i</i> -butanol	C <sub>4</sub> H <sub>10</sub> O	107.9	-108.2	0.803	8.5	0.552			
2-butanone	C <sub>4</sub> H <sub>8</sub> O	79.6	-86.3	0.805	25.6	0.327	0.51	200	105
<i>t</i> -butyl alcohol	C <sub>4</sub> H <sub>10</sub> O	82.2	25.5	0.786	M	0.389		100	41
carbon disulfide	CS <sub>2</sub>	46.3	-111.6	1.263	0.2	0.065	0.15	10	400
carbon tetrachloride	CCl <sub>4</sub>	76.7	-22.4	1.594	0.08	0.052	0.18	5	120
chlorobenzene	C <sub>6</sub> H <sub>5</sub> Cl	132	-45.6	1.106	0.05	0.188	0.30	10	12
chloroform	CHCl <sub>3</sub>	61.2	-63.5	1.498	0.8	0.259		10	210
cyclohexane	C <sub>6</sub> H <sub>12</sub>	80.7	6.6	0.779	0.005	0.006	0.04	100	104
cyclohexanol	C <sub>6</sub> H <sub>12</sub> O	161.1	25.2	0.962	4.2	0.509		50	1.2
cyclohexanone	C <sub>6</sub> H <sub>10</sub> O	155.6	-16.4	0.948	2.3	0.281		25	5
di- <i>n</i> -butylphthalate	C <sub>16</sub> H <sub>22</sub> O <sub>4</sub>	340	-35	1.049	0.0011	0.272			
1,1-dichloroethane	C <sub>2</sub> H <sub>4</sub> Cl <sub>2</sub>	57.3	-97.0	1.176	0.5	0.269		100	240
1,2-dichloroethane	C <sub>2</sub> H <sub>4</sub> Cl <sub>2</sub>	83.5	-35.4	1.235	0.87	0.327			

## AppB-2

<b>diethylamine</b>	C <sub>4</sub> H <sub>11</sub> N	56.3	-48	0.706	M	0.145	0.63	5	260
<b>diethylene glycol</b>	C <sub>4</sub> H <sub>10</sub> O <sub>3</sub>	245	-10	1.118	M	0.713			0.027
<b>diglyme</b>	C <sub>6</sub> H <sub>14</sub> O <sub>3</sub>	162	-64	0.945	M	0.244			
<b>dimethoxy-ethane (glyme)</b>	C <sub>4</sub> H <sub>10</sub> O <sub>2</sub>	85	-58	0.868	M	0.231			
<b>N,N-dimethyl-aniline</b>	C <sub>8</sub> H <sub>11</sub> N	194.2	2.4	0.956	0.14	0.179			
<b>dimethyl-formamide (DMF)</b>	C <sub>3</sub> H <sub>7</sub> NO	153	-61	0.944	M	0.386		10	3.5
<b>dimethyl phthalate</b>	C <sub>10</sub> H <sub>10</sub> O <sub>4</sub>	283.8	1	1.190	0.43	0.309			
<b>dimethyl sulfoxide (DMSO)</b>	C <sub>2</sub> H <sub>6</sub> OS	189	18.4	1.092	M	0.444	0.75		
<b>dioxane</b>	C <sub>4</sub> H <sub>8</sub> O <sub>2</sub>	101.1	11.8	1.033	M	0.164	0.56	20	41
<b>ethanol</b>	C <sub>2</sub> H <sub>6</sub> O	78.5	-114.1	0.789	M	0.654	0.88	100	59
<b>ether</b>	C <sub>4</sub> H <sub>10</sub> O	34.6	-116.3	0.713	7.5	0.117	0.38	400	587
<b>ethyl acetate</b>	C <sub>4</sub> H <sub>8</sub> O <sub>2</sub>	77	-83.6	0.894	8.7	0.228	0.58	400	97
<b>ethyl acetoacetate</b>	C <sub>6</sub> H <sub>10</sub> O <sub>3</sub>	180.4	-80	1.028	2.9	0.577			
<b>ethyl benzoate</b>	C <sub>9</sub> H <sub>10</sub> O <sub>2</sub>	213	-34.6	1.047	0.07	0.228			
<b>ethylene glycol</b>	C <sub>2</sub> H <sub>6</sub> O <sub>2</sub>	197	-13	1.115	M	0.790	1.11		
<b>glycerin</b>	C <sub>3</sub> H <sub>8</sub> O <sub>3</sub>	290	17.8	1.261	M	0.812			
<b>heptane</b>	C <sub>7</sub> H <sub>16</sub>	98	-90.6	0.684	0.0003	0.012		400	48
<b>hexane</b>	C <sub>6</sub> H <sub>14</sub>	69	-95	0.655	0.0014	0.009	0.01	50	160
<b>1-hexanol</b>	C <sub>6</sub> H <sub>14</sub> O	158	-46.7	0.814	0.59	0.559			
<b>methanol</b>	CH <sub>4</sub> O	64.6	-98	0.791	M	0.762	0.95	200	128
<b>methyl acetate</b>	C <sub>3</sub> H <sub>6</sub> O <sub>2</sub>	56.9	-98.1	0.933	24.4	0.253		200	220
<b>methyl <i>t</i>-butyl ether (MTBE)</b>	C <sub>5</sub> H <sub>12</sub> O	55.2	-109	0.741	4.8	0.124	0.20		
<b>methylene chloride</b>	CH <sub>2</sub> Cl <sub>2</sub>	39.8	-96.7	1.326	1.32	0.309	0.42	50	475
<b>1-octanol</b>	C <sub>8</sub> H <sub>18</sub> O	194.4	-15	0.827	0.096	0.537			
<b>pentane</b>	C <sub>5</sub> H <sub>12</sub>	36.1	-129.7	0.626	0.004	0.009	0.00	600	573
<b>1-pentanol</b>	C <sub>5</sub> H <sub>12</sub> O	138.0	-78.2	0.814	2.2	0.568			
<b>2-pentanol</b>	C <sub>5</sub> H <sub>12</sub> O	119.0	-50	0.810	4.5	0.488			
<b>3-pentanol</b>	C <sub>5</sub> H <sub>12</sub> O	115.3	-8	0.821	5.1	0.463			
<b>2-pentanone</b>	C <sub>5</sub> H <sub>10</sub> O	102.3	-76.9	0.809	4.3	0.321			
<b>3-pentanone</b>	C <sub>5</sub> H <sub>12</sub> O	101.7	-39.8	0.814	3.4	0.265		200	

<b>1-propanol</b>	C <sub>3</sub> H <sub>8</sub> O	97	-126	0.803	M	0.617	0.82		
<b>2-propanol</b>	C <sub>3</sub> H <sub>8</sub> O	82.4	-88.5	0.785	M	0.546	0.82	400	44
<b>pyridine</b>	C <sub>5</sub> H <sub>5</sub> N	115.5	-42	0.982	M	0.302	0.71	5	20
<b>tetrahydrofuran (THF)</b>	C <sub>4</sub> H <sub>8</sub> O	66	-108.4	0.886	30	0.207	0.57	200	200
<b>toluene</b>	C <sub>7</sub> H <sub>8</sub>	110.6	-93	0.867	0.05	0.099	0.29	50	29
<b>water</b>	H <sub>2</sub> O	100.00	0.00	0.998	M	1.000	>>1		
<b>water, heavy</b>	D <sub>2</sub> O	101.3	4	1.107	M	0.991			
<b>p-xylene</b>	C <sub>8</sub> H <sub>10</sub>	138.3	13.3	0.861	0.02	0.074	0.26	100	15

1 M = miscible.

2 The values for relative polarity are normalized from measurements of solvent shifts of absorption spectra and were extracted from Christian Reichardt, *Solvents and Solvent Effects in Organic Chemistry*, Wiley-VCH Publishers, 3rd ed., 2003.

3 Snyder's empirical eluant strength parameter for alumina. Extracted from Reichardt, page 495.

4 Threshold limits for exposure. Extracted from Reichardt, pages 501-502.

**TABLE 2**

<u>Solvent</u>	formula	boiling point (°C)	melting point (°C)	density (g/mL)	solubility in H <sub>2</sub> O <sup>1</sup> (g/100g)	relative polarity <sup>2</sup>	eluant strength <sup>3</sup>	threshold limits <sup>4</sup> (ppm)	vapor pressure 20°C (hPa)
cyclohexane	C <sub>6</sub> H <sub>12</sub>	80.7	6.6	0.779	0.005	<b>0.006</b>	0.04	100	104
pentane	C <sub>5</sub> H <sub>12</sub>	36.1	-129.7	0.626	0.0039	<b>0.009</b>	0.00	600	573
hexane	C <sub>6</sub> H <sub>14</sub>	69	-95	0.655	0.0014	<b>0.009</b>	0.01	50	160
heptane	C <sub>7</sub> H <sub>16</sub>	98	-90.6	0.684	0.0003	<b>0.012</b>		400	48
carbon tetrachloride	CCl <sub>4</sub>	76.7	-22.4	1.594	0.08	<b>0.052</b>	0.18	5	120
carbon disulfide	CS <sub>2</sub>	46.3	-111.6	1.263	0.2	<b>0.065</b>	0.15	10	400
p-xylene	C <sub>8</sub> H <sub>10</sub>	138.3	13.3	0.861	0.02	<b>0.074</b>	0.26	100	15
toluene	C <sub>7</sub> H <sub>8</sub>	110.6	-93	0.867	0.05	<b>0.099</b>	0.24	50	29
benzene	C <sub>6</sub> H <sub>6</sub>	80.1	5.5	0.879	0.18	<b>0.111</b>	0.32	0.5	101
ether	C <sub>4</sub> H <sub>10</sub> O	34.6	-116.3	0.713	7.5	<b>0.117</b>	0.38	400	587
methyl <i>t</i> -butyl ether (MTBE)	C <sub>5</sub> H <sub>12</sub> O	55.2	-109	0.741	4.8	<b>0.124</b>	0.20		
diethylamine	C <sub>4</sub> H <sub>11</sub> N	56.3	-48	0.706	M	<b>0.145</b>	0.63	5	260
dioxane	C <sub>4</sub> H <sub>8</sub> O <sub>2</sub>	101.1	11.8	1.033	M	<b>0.164</b>	0.56	20	41
N,N-dimethyl-aniline	C <sub>8</sub> H <sub>11</sub> N	194.2	2.4	0.956	0.14	<b>0.179</b>			
chlorobenzene	C <sub>6</sub> H <sub>5</sub> Cl	132	-45.6	1.106	0.05	<b>0.188</b>	0.30	10	12
anisole	C <sub>7</sub> H <sub>8</sub> O	153.7	-37.5	0.996	0.10	<b>0.198</b>			
tetrahydrofuran (THF)	C <sub>4</sub> H <sub>8</sub> O	66	-108.4	0.886	30	<b>0.207</b>	0.57	200	200

## AppB-4

ethyl acetate	C <sub>4</sub> H <sub>8</sub> O <sub>2</sub>	77	-83.6	0.894	8.7	<b>0.228</b>	0.57	400	97
ethyl benzoate	C <sub>9</sub> H <sub>10</sub> O <sub>2</sub>	213	-34.6	1.047	0.07	<b>0.228</b>			
dimethoxy-ethane (glyme)	C <sub>4</sub> H <sub>10</sub> O <sub>2</sub>	85	-58	0.868	M	<b>0.231</b>			
diglyme	C <sub>6</sub> H <sub>14</sub> O <sub>3</sub>	162	-64	0.945	M	<b>0.244</b>			
methyl acetate	C <sub>3</sub> H <sub>6</sub> O <sub>2</sub>	56.9	-98.1	0.933	24.4	<b>0.253</b>		200	220
chloroform	CHCl <sub>3</sub>	61.2	-63.5	1.498	0.8	<b>0.259</b>	0.40	10	210
3-pentanone	C <sub>5</sub> H <sub>12</sub> O	101.7	-39.8	0.814	3.4	<b>0.265</b>		200	
1,1-dichloroethane	C <sub>2</sub> H <sub>4</sub> Cl <sub>2</sub>	57.3	-97.0	1.176	0.5	<b>0.269</b>		100	240
di-n-butyl phthalate	C <sub>16</sub> H <sub>22</sub> O <sub>4</sub>	340	-35	1.049	0.0011	<b>0.272</b>			
cyclohexanone	C <sub>6</sub> H <sub>10</sub> O	155.6	-16.4	0.948	2.3	<b>0.281</b>		25	5
pyridine	C <sub>5</sub> H <sub>5</sub> N	115.5	-42	0.982	M	<b>0.302</b>	0.71	5	20
dimethylphthalate	C <sub>10</sub> H <sub>10</sub> O <sub>4</sub>	283.8	1	1.190	0.43	<b>0.309</b>			
methylene chloride	CH <sub>2</sub> Cl <sub>2</sub>	39.8	-96.7	1.326	1.32	<b>0.309</b>	0.42	50	475
2-pentanone	C <sub>5</sub> H <sub>10</sub> O	102.3	-76.9	0.809	4.3	<b>0.321</b>			
2-butanone	C <sub>4</sub> H <sub>8</sub> O	79.6	-86.3	0.805	25.6	<b>0.327</b>	0.51	200	105
1,2-dichloroethane	C <sub>2</sub> H <sub>4</sub> Cl <sub>2</sub>	83.5	-35.4	1.235	0.87	<b>0.327</b>			
benzonitrile	C <sub>7</sub> H <sub>5</sub> N	205	-13	0.996	0.2	<b>0.333</b>		10	12
acetone	C <sub>3</sub> H <sub>6</sub> O	56.2	-94.3	0.786	M	<b>0.355</b>	0.56	500	240
dimethyl-formamide (DMF)	C <sub>3</sub> H <sub>7</sub> NO	153	-61	0.944	M	<b>0.386</b>		10	3.5
<i>t</i> -butyl alcohol	C <sub>4</sub> H <sub>10</sub> O	82.2	25.5	0.786	M	<b>0.389</b>		100	41
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dimethyl sulfoxide (DMSO)	C <sub>2</sub> H <sub>6</sub> OS	189	18.4	1.092	M	<b>0.444</b>	0.75		
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2-pentanol	C <sub>5</sub> H <sub>12</sub> O	119.0	-50	0.810	4.5	<b>0.488</b>			
2-butanol	C <sub>4</sub> H <sub>10</sub> O	99.5	-114.7	0.808	18.1	<b>0.506</b>		100	
cyclohexanol	C <sub>6</sub> H <sub>12</sub> O	161.1	25.2	0.962	4.2	<b>0.509</b>		50	1.2
1-octanol	C <sub>8</sub> H <sub>18</sub> O	194.4	-15	0.827	0.096	<b>0.537</b>			
2-propanol	C <sub>3</sub> H <sub>8</sub> O	82.4	-88.5	0.785	M	<b>0.546</b>	0.82	400	44
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acetyl acetone	C <sub>5</sub> H <sub>8</sub> O <sub>2</sub>	140.4	-23	0.975	16	<b>0.571</b>			
ethyl acetoacetate	C <sub>6</sub> H <sub>10</sub> O <sub>3</sub>	180.4	-80	1.028	2.9	<b>0.577</b>			
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benzyl alcohol	C <sub>7</sub> H <sub>8</sub> O	205.4	-15.3	1.042	3.5	<b>0.608</b>			
1-propanol	C <sub>3</sub> H <sub>8</sub> O	97	-126	0.803	M	<b>0.617</b>	0.82		
acetic acid	C <sub>2</sub> H <sub>4</sub> O <sub>2</sub>	118	16.6	1.049	M	<b>0.648</b>	>1	10	15.3
2-aminoethanol	C <sub>2</sub> H <sub>7</sub> NO	170.9	10.5	1.018	M	<b>0.651</b>		3	0.53
ethanol	C <sub>2</sub> H <sub>6</sub> O	78.5	-114.1	0.789	M	<b>0.654</b>	0.88	1000	59
diethylene glycol	C <sub>4</sub> H <sub>10</sub> O <sub>3</sub>	245	-10	1.118	M	<b>0.713</b>			0.027
methanol	CH <sub>4</sub> O	64.6	-98	0.791	M	<b>0.762</b>	0.95	200	128
ethylene glycol	C <sub>2</sub> H <sub>6</sub> O <sub>2</sub>	197	-13	1.115	M	<b>0.790</b>	1.11		
glycerin	C <sub>3</sub> H <sub>8</sub> O <sub>3</sub>	290	17.8	1.261	M	<b>0.812</b>			
water, heavy	D <sub>2</sub> O	101.3	4	1.107	M	<b>0.991</b>			
water	H <sub>2</sub> O	100.00	0.00	0.998	M	<b>1.000</b>	>>1		